

# **MATERIAL SAFETY DATA SHEET**

#### **SECTION 1 - GENERAL INFORMATION**

MANUFACTURER'S CAMDENBOSS LTD	EMERGENCY TELEPHONE NO.: 01638 716 101
ADDRESS: 60 James Carter Rd, Mildenhall, Suffolk, IP28	OTHER INFORMATION CALLS: 01638 716 101
PERSON RESPONSIBLE FOR PREPARATION Shouzhong Yi, Safety, Health & Environmental Affairs Manager	Revised Date: JUNE 30, 2009

#### **SECTION 2 - COMPOSITION/INFORMATION ON INGREDIENTS**

C.A.S.	PRINCIPAL HAZARDOUS COMPONENT(S) (chemical & common name(s)	Hazard Category	% Weight	ACGIH TLV - mg/m <sup>3</sup>	OSHA PEL/TWA - mg/m <sup>3</sup>
7439-92-1	Lead/Lead Oxide (Litharge)/Lead Sulfate	Acute-Chronic	60-70	0.05 mg/m <sup>3</sup>	0.05 mg/m <sup>3</sup>
7440-70-2	Calcium (lead calcium alloy)	Reactive	<0.15	Not Established	Not Established
7440-31-5	Tin	Chronic	<1	2	2
7440-38-2	Arsenic (inorganic)	Acute-Chronic	<1	0.01	0.01
7664-93-9 Sulfuric Acid (Battery Electrolyte)		Reactive-Oxidizer Acute -Chronic	10-15	1.0	1.0
Not applicable	Inert Ingredients	Not applicable	<6	Not Applicable	Not Applicable

Note: PEL's for Individual states may differ from OSHA's PEL's. Check with local authorities for the applicable state PEL's. OSHA – Occupational Safety and Health Administration: ACGIH – American Conference of Governmental Industrial Hydienists: NIOSH – National

OSHA – Occupational Safety and Health Administration; ACGIH – American Conference of Governmental Industrial Hygienists; NIOSH – National Institute for Occupational Safety and Health.

COMMON NAME: (Used on label) Valve Regulated Lead-acid Battery

(Trade Name & Synonyms) VRB, VRLA, SLAB, Recombinant lead acid: RG, GPL, AGM, PVX or FD Series, D8565 series

Chemical Family: Toxic and Corrosive Material Mixture

Chemical Name: Bat Formula: Lead/Acid

# Name: Battery, Storage, Lead Acid, Valve Regulated SECTION 3 -- HAZARD IDENTIFICATION

Signs and Symptoms of Exposure	1. Acute Hazards	Do not open battery. Avoid contact with internal components. Internal components include lead and absorbed electrolyte. Electrolyte - Electrolyte is corrosive and contact may cause skin irritation and chemical burns. Electrolyte causes severe irritation and burns of eyes, nose and throat. Ingestion can cause severe burns and vomiting. Lead - Direct skin or eye contact may cause local irritation. Inhalation or ingestion of lead dust or fumes may result in headache, nausea, vomiting, abdominal spasms, fatigue, sleep disturbances, weight loss, anemia and leg, arm and joint pain.					
	2. Subchronic and Chronic Health Effects	teeth, chronic eye i Lead - Prolonged e taste, insomnia, wr excessive exposure <u>California Propos</u> i	lyte - Repeated contact with electrolyte causes irritation and skin burns. Repeated exposure to mist may cause erosion or chronic eye irritation and/or chronic inflammation of the nose, throat and lungs. Prolonged exposure may cause central nervous system damage, gastrointestinal disturbances, anemia, irritability, metalli nsomnia, wrist-drop, kidney dysfunction and reproductive system disturbances. Pregnant women should be protected fro ive exposure to prevent lead from crossing the placental barrier and causing infant neurological disorders. <u>rnia Proposition 65 Warning</u> : Battery posts, terminals, and related accessories contain lead and lead compounds, chem to the State of California to cause cancer and reproductive harm, and during charging, strong inorganic acid mists contair				
Medical Conditions Generally Aggravated by Exposure		ernal components if ba		n persons with the following med			
Routes of Entry	Inhalation - YES Ingestion - YES			Eye Contact- YES			
Chemical(s) Listed as Carcinogen or potential Carcinogen			Proposition 65 - YES	National Toxicology Program	n - I.A.R.C. Monographs - YE	O.S.H.A NO S	

#### **SECTION 4 - FIRST AID MEASURES**

Emergency and First Aid	Contact with internal components if battery is opened/broken.
Procedures	
1. Inhalation	Remove to fresh air and provide medical oxygen/CPR if needed. Obtain medical attention.
2. Eyes	Immediately flush with water for at least 15 minutes, hold eyelids open. Obtain medical attention.
-	
3. Skin	Flush contacted area with large amounts of water for at least 15 minutes. Remove contaminated clothing and obtain medical attention if
	necessary.
4. Ingestion	Do not induce vomiting. If conscious drink large amounts of water/milk. Obtain medical attention. Never give anything by mouth to an
-	unconscious person.

#### **SECTION 5 - FIREFIGHTING MEASURES**

Flash Point – Not Applicable	Flammable Limits in Air % by Volume: Not Applicable	Extinguishing Media – Class ABC, CO <sub>2</sub> , Halon	Auto-Ignition 675°F (polypropylene) Temperature			
Special Fire Fighting Procedures	Lead/acid batteries do not burn, or burn with difficulty. Do not use water on fires where molten metal is present. Extinguish fire with agent suitable for surrounding combustible materials. Cool exterior of battery if exposed to fire to prevent rupture. The acid mist and vapors generated by heat or fire are corrosive. Use NIOSH approved self-contained breathing apparatus (SCBA) and full protective equipment operated in positive-pressure mode.					
Unusual Fire and Explosion Hazards	Sulfuric acid vapors are generated upon overcharge and polypropylene case failure. Use adequate ventilation. Avoid open flames/sparks/other sources of ignition near battery.					

### **SECTION 6 - ACCIDENTAL RELEASE MEASURES**

Procedures for Cleanup. Avoid contact with any spilled material. Contain spill, isolate hazard area, and deny entry. Limit site access to emergency responders. Neutralize with sodium bicarbonate, soda ash, lime or other neutralizing agent. Place battery in suitable container for disposal. Dispose of contaminated material in accordance with applicable local, state and federal regulations. Sodium bicarbonate, soda ash, sand, lime or other neutralizing agent should be kept on-site for spill remediation.

Personal Precautions: Acid resistant aprons, boots and protective clothing. ANSI approved safety glasses with side shields/face shield recommended.

Environmental Precautions: Lead and its compounds and sulfuric acid can pose a severe threat to the environment. Contamination of water, soil and air should be prevented.

# **SECTION 7 - HANDLING AND STORAGE**

Precautions to be Taken in Handling and Storage	Store away from reactive materials, open flames and sources of ignition as defined in Section 10 – Stability and Reactivity Data. Store batteries in cool, dry, well-ventilated areas. Batteries should be stored under roof for protection against adverse weather conditions. Avoid				
	image to containers.				
Other Precautions	GOOD PERSONAL HYGIENE AND WORK PRACTICES ARE MANDATORY. Refrain from eating, drinking or smoking in work areas. Thoroughly wash hands, face, neck and arms, before eating, drinking and smoking. Work clothes and equipment should remain in designated lead contaminated areas, and never taken home or laundered with personal clothing. Wash soiled clothing, work clothes and equipment before reuse.				

#### **SECTION 8 - EXPOSURE CONTROLS AND PERSONAL PROTECTION**

Respiratory	None required under normal conditions. Acid/gas NIOSH approved respirator is required when the PEL is exceeded or employee experiences							
Protection	respiratory irritation.							
(Specify Type)								
Ventilation	Store and handle in dry ventilated Local When PEL is exceeded. Mechanical Not Applicable					Not Applicable		
	area.	Exhaust	Exhaust					
Protective	Wear rubber or plastic acid resistant	gloves.	Eye Protection	ANSI app	roved safety glass	es with side shields/face shield recommended		
Gloves			-					
Other Protective	Safety shower and eyewash.							
Clothing or								
Equipment								

## **SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES**

		plicable			Melting Point: >320°F (polypropylene)			
	Pressure			Gravity				
Percent Volatile Not Applic	able	Vapor	Hydrogen:	0.069	(Air =1)		Evaporation	Not applicable
By Volume		Density	Electrolyte:	3.4 @ STP	(Air = 1)		Rate	
Solubility 100% soluble (electrolyte)			Reactivity in Water Electrolyte – Water Reactive (1)					
In water						-		
Appearance and Odor:	Battery: Co-polymer	polypropylene,	, solid; may be	contained w	ithin an outer/	casing of alur	minum or steel. Cas	se has metal terminals.
Lead: Gray, metallic, solid; brown/grey oxide					. 15			
Electrolyte: Odorless, liquid absorbed in glass mat			material.					
	No apparent odor.		-					

# SECTION 10 - STABILITY AND REACTIVITY

		_
Stability: Stable	Conditions to Avoid: Avoid overcharging and smoking, or sparks near battery surface. High temperatures-cases decompose at >320°F.	
Incompatibility	Sparks, open flames, keep battery away from strong oxidizers.	
(Materials to Avoid)		- 1
Hazardous	Combustion can produce carbon dioxide and carbon monoxide.	
Decomposition Products		
Hazardous	Hazardous Polymerization has not been reported.	Т
Polymerization		- 1

# SECTION 11 - TOXICOLOGICAL INFORMATION

GENERAL: The primary routes of exposure to lead are ingestion or inhalation of dust and fumes.

#### ACUTE:

INHALATION/INGESTION: Exposure to lead and its compounds may cause headache, nausea, vomiting, abdominal spasms, fatigue, sleep disturbances, weight loss, anemia, and pain in the legs, arms and joints. Kidney damage, as well as anemia, can occur from acute exposure.

#### CHRONIC:

INHALATION/INGESTION: Prolonged exposure to lead and its compounds may produce many of the symptoms of short-term exposure and may also cause central nervous system damage, gastrointestinal disturbances, anemia, and wrist drop. Symptoms of central nervous system damage include fatigue, headaches, tremors, hypertension, hallucination, convulsions and delirium. Kidney dysfunction and possible injury has also been associated with chronic lead poisoning. Chronic over-exposure to lead has been implicated as a causative agent for the impairment of male and female reproductive capacity, but there is at present, no substantiation of the implication. Pregnant women should be protected from excessive exposure. Lead can cross the placental barrier and unborn children may suffer neurological damage or developmental problems due to excessive lead exposure in pregnant women.

#### **SECTION 12 - ECOLOGICAL INFORMATION**

In most surface water and groundwater, lead forms compounds with anions such as hydroxides, carbonates, sulfates, and phosphates, and precipitates out of the water column. Lead may occur as sorbed ions or surface coatings on sediment mineral particles or may be carried in colloidal particles in surface water. Most lead is strongly retained in soil, resulting in little mobility. Lead may be immobilized by ion exchange with hydrous oxides or clays or by chelation with humic or fulvic acids in the soil. Lead (dissolved phase) is bioaccumulated by plants and animals, both aquatic and terrestrial.

#### **SECTION 13 - DISPOSAL CONSIDERATIONS**

Lead-acid batteries are completely recyclable. Return whole scrap batteries to distributor, manufacturer or lead smelter for recycling. For information on returning batteries for recycling call 01638 716 101. For neutralized spills, place residue in acid-resistant containers with sorbent material, sand or earth and dispose of in accordance with local, state and federal regulations for acid and lead compounds. Contact local and/or state environmental officials regarding disposal information.

#### **SECTION 14 - TRANSPORT INFORMATION**

All CamdenBoss' AGM, CP, FM, CL series and CTA series are valve regulated lead acid (VRLA) batteries.

CamdenBoss' VRLA batteries have passed vibration, pressure differential and free flowing acid tests under CFR 49 173. 159(d) and the substance is not restricted to IMO IMDG code according to special provision 238. They are protected from short circuits and labeled "Non-Spillable." CamdenBoss' VRLA batteries are exempt fromDOT Hazardous Material Regulations and IMDG Dangerous Goods Regulations.

Note: The shipper has the option of shipping the batteries Hazmat regulated under UN2800. Additional labeling and paperwork would be required. See CFR 49 and IATA Dangerous Goods Regulations for more information.

U.S. DOT PROPER SHIPPING NAME U.S. DOT HAZARD CLASS: 8	E: Batteries, wet, non-spillable	e	
U.S. DOT ID NUMBER: UN2800		OR	Excepted from the requirements because batteries have passed the Vibration and
U.S. DOT PACKING GROUP: III			Pressure Differential performance tests, and ruptured case test for Nonspillable designation.
U.S. DOT LABEL: CORROSIVE			
IMO PROPER SHIPPING NAME: Bat IMO U.N. CLASS: 8 IMO U.N. NUMBER: UN 2800 IMO LABEL: CORROSIVE IMO VESSEL STOWAGE: A	tteries, wet, non-spillable	Ems # - F-/	A, S-B
IATA PROPER SHIPPING NAME: Ba IATA U.N. CLASS: 8 hen		OR	Excepted from the requirements because batteries have passed the vibration and pressure differential performance tests, and ruptured case test for nonspillable designation. And,
IATA U.N. NUMBER: UN 2800 IATA LABEL: CORROSIVE ERG Code – 8L		packaged f	for transport, the terminals are protected from short circuit.

#### SECTION 15 - REGULATORY INFORMATION

U.S. HAZARDOUS UNDER HAZARD COMMUNICATION STANDAR	LEAD - YES ARSENIC – YES SULFURIC ACID – YES	
INGREDIENTS LISTED ON TSCA INVENTORY:	YES	
CERCLA SECTION 304 HAZARDOUS SUBSTANCES:	LEAD – YES ARSENIC – YES SULFURIC ACID – YES	RQ: N/A* RQ: 1 POUND RQ: 1000 POUNDS

\* RQ: REPORTING NOT REQUIRED WHEN DIAMETER OF THE PIECES OF SOLID METAL RELEASED IS EQUAL TO OR EXCEEDS 100 μm (micrometers).

EPCRA SECTION 302 EXTREMELY HAZARDOUS SUBSTANCE:

EPCRA SECTION 313 TOXIC RELEASE INVENTORY:

SULFURIC ACID - YES

LEAD – CAS NO: 7439-92-1 ARSENIC – CAS NO: 7440-38-2 SULFURIC ACID – CAS NO: 7664-93-9